

## TWO AURONE GLYCOSIDES FROM THE FLOWERS OF *PTEROCARPUS MARSUPIUM*

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**Key Word Index**—*Pterocarpus marsupium*, Leguminosae, 4,6,4'-trihydroxyaurone 6-O-rhamnopyranoside; 4,6,4'-trihydroxy-7-methylaurone 4-O-rhamnopyranoside

**Abstract**—Two new aurone glycosides, 4,6,4'-trihydroxyaurone 6-O-rhamnopyranoside and 4,6,4'-trihydroxy-7-methylaurone 4-O-rhamnopyranoside have been isolated and identified from the flowers of *Pterocarpus marsupium*.

### INTRODUCTION

Species of *Pterocarpus* are known to be rich in iso-flavonoids [1] and terpenoid [2] derivatives. A flavonoid fraction of *P. marsupium* bark was found to effectively reverse the alloxan-induced changes in the blood sugar level and the beta cell population in the pancreas [3]. From the ethanolic extract of the flowers of *P. marsupium* two new aurone glycosides have been isolated and identified. This is the first report of aurones from *Pterocarpus* species.

### RESULTS AND DISCUSSION

Compound 1, the yellow pigment analysed for  $C_{21}H_{20}O_9$ , mp 178°. It was found to be glycosidic in nature [4]. On acid hydrolysis (7%  $H_2SO_4$ ) it gave an aglycone and rhamnose, identified by co-chromatography with an authentic sample and by  $^1H$  NMR spectral analysis of the glycoside (a doublet at  $\delta$  1.20 corresponding to three protons of rhamnosyl -Me group, broad signal at  $\delta$  3.5–3.82 for four sugar protons and a singlet at  $\delta$  4.2 due to C-1'' proton of rhamnose).

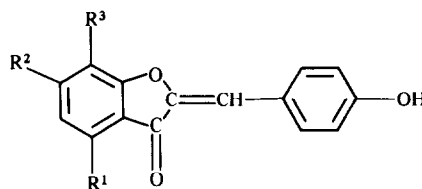
The aglycone,  $C_{15}H_{10}O_5$ , was characterized as an aurone on the basis of colour reactions [5] and UV spectral data [6].  $^1H$  NMR studies showed six aromatic protons suggesting a trisubstituted aurone. A multiplet at  $\delta$  7.7–7.9 (2H) was due to C-2' and -6' and multiplet at  $\delta$  6.8–7.0 (2H) for C-3' and -5' protons. Two doublets at  $\delta$  6.1 and 6.3 ( $J = 2$  Hz) were due to C-5 and C-7 protons, respectively. There was a singlet at  $\delta$  6.67 for benzylic proton (=CH-) [7]. On acetylation it gave a triacetate, mp 62°, showing the presence of three hydroxyl groups. The positions of hydroxyls in the glycoside were confirmed by UV shifts [6]. The presence of a free -OH at C-4' position was confirmed by a large bathochromic shift in  $\lambda_{max}^{MeOH}$  (ca 70 nm) with the addition of sodium methoxide and (68 nm) with sodium acetate. A large bathochromic shift in the  $\lambda_{max}^{MeOH}$  (ca 60 nm) with the addition of aluminium chloride confirmed the presence of a free -OH at position C-4 and the compound as 6-rhamnoside [8].

In the mass fragmentation pattern [7] two fragments at  $m/z$  152 and  $m/z$  118 showed that one hydroxyl group was

present in the B ring and the other in A ring. Thus the structure of 1 is confirmed; this a new glycoside but the aglycone has been reported from flowers of *Limonium* [9].

Compound 2, analysed for  $C_{22}H_{22}O_9$ , mp 168°, was found to be glycosidic in nature [4] and on hydrolysis it gave rhamnose (co-PC) and an aglycone which was shown to be an aurone by its colour reactions [5] and UV spectrum [6].  $^1H$  NMR spectrum of the glycoside showed a doublet at  $\delta$  1.20 corresponding to three protons of rhamnosyl -Me group.

$^1H$  NMR studies of the aglycone showed five aromatic protons suggesting a tetrasubstituted nuclei. The multiplet at  $\delta$  7.8–7.9 (2H) was attributed to C-2' and -6' and a multiplet at  $\delta$  6.88–7.10 (2H) to C-3' and -5' protons. A singlet at  $\delta$  6.25 (1H) was due to C-5 proton. The aglycone  $C_{16}H_{12}O_5$ , analysed for three hydroxyls (triacetate, IR 3374  $cm^{-1}$ ) and a  $^1H$  NMR signal at  $\delta$  1.45 corresponding to 3H of the methyl group. The two free -OH groups in the glycoside were shown to be at positions-4' and -6' by UV spectral shifts (bathochromic shift of 45 and 46 nm of band I with sodium methoxide and sodium acetate, respectively) [6]. No change in the  $\lambda_{max}^{MeOH}$  with the addition of aluminium chloride confirmed that this compound was a 4-rhamnoside [8]. C-7 position of -Me group was confirmed by  $^1H$  NMR studies.



- 1  $R^3 = H$ ,  $R^1 = OH$ ,  $R^2 = O-rha$   
 2  $R^3 = Me$ ,  $R^2 = OH$ ,  $R^1 = O-rha$

Mass spectral data showed a molecular ion peak at 430. Two fragments at  $m/z$  166 and 118 showed that one -OH was present in the B ring and a -OH and -Me were present in the A ring. Thus **2** is a new compound, namely, 4,6,4'-trihydroxy-7-methylaurone 4-O-rhamnopyranoside. The aglycones of **1** and **2** were further confirmed by synthesis. Aglycone of **1** was synthesized by condensing 4,6-dihydroxycoumaranone with *p*-hydroxybenzaldehyde [10]. The aglycone of **2** was prepared from 4,6-dihydroxy-7-methylcoumaranone and *p*-hydroxybenzaldehyde.

#### EXPERIMENTAL

The flowers of *Pterocarpus marsupium* were identified by the botanical survey of India. The air-dried and crushed flowers were extracted with boiling EtOH. The concd extract (150 ml) fractionated into petrol and  $C_6H_6$ . The remaining mother liquor was concd and chromatographed over a silica gel column. Compound **1** was extracted from  $CHCl_3$ -EtOAc fractions and compound **2** from EtOAc fraction. The compounds were crystallized from MeOH. Purity of the compounds was checked by PC and TLC.

Compound **1**, mp 178°,  $C_{21}H_{20}O_9$  (Found C, 60.62, H 4.68. Calc. for  $C_{21}H_{20}O_9$ , C 60.57, H 4.80%) UV  $\lambda_{max}^{MeOH}$  nm 225, 245 (sh), 345 (sh), 392, +  $AlCl_3$  225, 245 (sh), 348 (sh), 452, + NaOMe 227, 247 (sh), 346 (sh), 462, + NaOAc 227, 248 (sh), 350, 460. IR  $\nu_{max}^{KBr}$   $cm^{-1}$  3368 (-OH), 1632 (>C=O), 730, 680.  $^1H$  NMR (60 MHz,  $CDCl_3$ )  $\delta$  1.2 (3H, d, rhamnose -Me), 4.2 (1H, s, H-1''), 3.55-3.8 (4H, br, sugar protons), 7.7-7.9 (2H, m, H-2' and -6'), 6.1 (1H, d,  $J$  = 2 Hz, H-5), 6.3 (1H, d,  $J$  = 2 Hz, H-7), 6.74 (1H, s, =CH-), 6.8-7.0 (2H, m, H-3' and -5'). MS (70 eV)  $m/z$ : 416 ( $M^+$ ), 270 (100), 269, 253, 242, 152, 124, 118. Acid hydrolysis (7%  $H_2SO_4$ ) of **1** yielded an aglycone ( $R_f$  0.75, solvent *n*-BuOH-HOAc- $H_2O$ , 4:1:5, spray  $I_2$  vapours) and rhamnose. Aglycone of **1**, mp 208°,  $C_{15}H_{10}O_5$  (Found C, 67.1, H 3.5, Calc C, 66.67, H 3.7%)  $^1H$  NMR (60 MHz,  $CDCl_3$ )  $\delta$ : 6.15 (1H, d, H-5), 6.36 (1H, d, H-7), 6.75 (1H, s, =CH-), 7.7-7.9 (2H, m, H-2' and -6'), 6.8-7.0 (2H, m, H-3' and -5'). Acetate (pyridine- $Ac_2O$ , 24 hr at room temp.) mp 62° (Found C, 63.58, H 4.09, acetyl 32.59). Requires C, 63.63, H 4.04, acetyl 32.57%. Compound **2**, mp 168°,  $C_{22}H_{22}O_9$  (Found C, 60.92, H 5.20. Calc for  $C_{22}H_{22}O_9$ , C 61.39, H 5.11%) UV  $\lambda_{max}^{MeOH}$  nm 220, 260, 350 (sh), 395, + NaOMe 225, 262 (sh), 360 (sh), 440, + NaOAc 222 (sh), 265 (sh), 359 (sh), 441, +  $AlCl_3$  222, 260, 352 (sh), 395. IR  $\nu_{max}^{KBr}$   $cm^{-1}$  3374 (-OH), 1638 (>C=O), 722, 668.  $^1H$  NMR (60 MHz,  $CDCl_3$ )  $\delta$  1.2 (3H, d, rhamnose -Me), 4.3 (1H, s, H-1''), 3.4-3.6 (4H, br, sugar protons), 7.8-7.9 (2H, m, H-2' and -6'), 6.88-7.1 (2H, m, H-3' and -5'), 6.25 (1H, s, H-5), 1.45 (3H, s, -Me), 6.65 (1H, s, =CH-) MS (70 eV)  $m/z$ : 430 ( $M^+$ ), 280 (100), 166, 138, 118. Acid hydrolysis of **2** with 7%  $H_2SO_4$  gave an aglycone ( $R_f$  0.71, solvent *n*-BuOH-HOAc- $H_2O$  4:1:5) and rhamnose. Aglycone of **2**, mp 160°,  $C_{16}H_{12}O_5$  (Found C, 68.2, H 4.10, Calc for  $C_{16}H_{12}O_5$ , C, 67.6, H 4.2%) Acetate (pyridine- $Ac_2O$ , 24 hr at room temp.) mp 92° (Found C, 64.31, H 4.42, acetyl 31.48, Calc C 64.39, acetyl 31.46%)

*Synthesis of 4,6,4'-trihydroxy-7-methylaurone* 2,4,6-trihydroxytoluene. To phloroglucinaldehyde [11] (1g) in HOAc (100  $cm^3$ ) Pd/C (5%, 1.5 g) was added and the mixture shaken in an atmosphere of pure  $H_2$ . Solution was filtered and catalyst washed with HOAc. From the extract HOAc was removed under red pres.  $H_2O$  was added and the product was extracted in  $Et_2O$ . The  $Et_2O$  extract was washed with  $NaHCO_3$  and then extracted with  $Na_2CO_3$  soln. Acidification of the carbonate solution gave crystalline needles, mp 52°.

*4,6-dihydroxy-7-methyl coumaranone* Dry HCl gas was passed for 3.5 hr into a stirred mixture of trihydroxytoluene (80 g), 50 g chloroacetonitrile, 20 g powdered dry  $ZnCl_2$  and 500 ml of dry  $Et_2O$ . A second phase oily at first and later solid, separated. The mix. was left overnight, the solid separated and washed with fresh  $Et_2O$ . The yellow solid was dissolved in 400 ml ice  $H_2O$  and the solution heated under reflux for 1 hr. The solid which separated on cooling was boiled for 5 hr in a solution of 100 g KOAc in 20 ml of water. The hot solution was treated with decolorizing carbon and filtered. The product was charged over a silica gel column. From the ethyl acetate fraction yellow coloured needles were obtained, mp 140-145° (dec).

*Synthesis of aurone* Equimolecular amounts (100 mg) of 4,6-dihydroxy-7-methylcoumaranone and *p*-hydroxybenzaldehyde were dissolved in 5 ml HOAc and 0.2 ml conc. HCl added [10]. After 4 hr at room temp., the solution was poured into water and the product collected and recrystallized from aqueous EtOH. mp 160°.

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#### REFERENCES

1. Mitra, J. and Joshi, T. (1983) *Phytochemistry* **22**, 2328.
2. Seshadri, T. R. (1972) *Phytochemistry* **11**, 881.
3. Chakravarty, B. K., Gupta, S., Gambhir, S. S. and Gode, K. D. (1980) *Indian J. Pharmacol.* **12(a)**, 123.
4. Molisch, H. (1886) *Monatsch. Chem.*, **7**, 108.
5. Jurd, Leonard (1962), *The Chemistry of Flavonoid Compounds* Geissman, T. A., (ed.), p. 82 Pergamon Press, New York.
6. Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) *The Systematic Identification of Flavonoids* Springer, Berlin.
7. Nascimento, M. C. D., Dias, R. L. D. V. and Mors, W. B. (1976) *Phytochemistry* **15**, 1153.
8. Geissman, T. A. and Harborne, J. B. (1956) *J. Am. Chem. Soc.* **78**, 832.
9. Asen, S. and Plimma, J. R. (1972) *Phytochemistry* **11**, 2601.
10. Geissman, T. A. and Harborne, J. B. (1955) *J. Am. Chem. Soc.* **77**, 4622.
11. Malkin, T. et al. (1931) *J. Am. Chem. Soc.* **53**, 239.